

## Research Article

# Use of Plackett–Burman Statistical Design to Study Effect of Formulation Variables on the Release of Drug from Hot Melt Sustained Release Extrudates

Satishkumar P. Jain,<sup>1</sup> Pirthi Pal Singh,<sup>1</sup> Sharad Javeer,<sup>1</sup> and Purnima D. Amin<sup>1,2</sup>

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**Abstract.** The present paper was focused on exploiting Plackett–Burman design to screen the effect of nine factors—poly (ethylene oxide) molecular weight ( $X_1$ ), poly (ethylene oxide) amount ( $X_2$ ), ethylcellulose amount ( $X_4$ ), drug solubility ( $X_5$ ), drug amount ( $X_6$ ), sodium chloride amount ( $X_7$ ), citric acid amount ( $X_8$ ), polyethylene glycol amount ( $X_9$ ), and glycerin amount ( $X_{11}$ ) on the release of drugs from the extended release extrudates, i.e., release rate and release mechanism. The experiments were carried out according to a nine-factor 12-run statistical model and subjected to an 8-h dissolution study in phosphate buffer pH 6.8. The significance of the model was indicated by the ANOVA and the residual analysis. Poly (ethylene oxide) amount, ethylcellulose amount and drug solubility had significant effect on the T90 values whereas poly (ethylene oxide) amount and ethylcellulose amount had significant effect on the  $n$  value.

**KEY WORDS:** ethylcellulose; extended release; hot melt extrusion; Plackett–Burman screening; poly (ethylene oxide).

## INTRODUCTION

Hot melt extrusion (HME) is a technique in which during extrusion, a polymer melt is pumped through a shaping die and formed into a profile. This profile can be a plate, a film, a tube, or have any shape of its cross section (1,2). HME technology has been exploited in polymer industries since the 1930s (3). Since then, it has been extensively used in polymer (4), food (5,6), chemical (7), rubber (8), and metal industries (9). In pharmaceutical industries, this technology is exploited for the manufacturing of pellets (10,11), solid dispersion (12–14), topical dosage forms (15), powder coating (16), gastro retentive dosage forms (17), tablets (18), sustained release oral dosage forms (19,20), and ophthalmic inserts (21).

Polymer choice is the most critical factor to obtain the desired drug release profile during formulation development of hot melt extrudates. Most commonly, the hot melt extrudates are comprised of drug with one or more functional excipients (22). Polymer, a deformable carrier, is the most essential component of HME that carries the poor thermo-plastic drug(s). The selection of the polymer as well as the drug to polymer ratio exhibits significant effect on the release profile of the dosage form (23). Hydrophilic polymers are most widely used for the development of extended release products. However, their use for controlling the release of

highly water-soluble drug is restricted due to rapid diffusion of the dissolved drug through the hydrophilic gel layer (24). Literature reports the use of combinations of hydrophilic and hydrophobic matrices as the preferred choice for the preparation of extended release dosage forms (24). Plasticizers such as polyethylene glycol and glycerin, the second most important component of HME, are used to improve the processability of polymers by increasing their flexibility and reducing extrusion temperature (25,26). Few drugs have also reported as potential plasticizers (27). Recently, drug release modifying agents such as sodium chloride and citric acid have been reported in the literature (25–28). The mechanism of drug release modifying agents is varied and most often it is linked to increase in drug release rate by diffusion, erosion, or creating channels. Citric acid also promotes the thermal processability and matrix integrity by plasticization of polymer (25).

The impact of various factors like polymer concentration, drug loading, drug solubility, diluent, and ionic concentration on the release of the drug from the extended release formulations has been reported in the literature (29–31). In the development of extended release formulations, it becomes difficult to ascertain factors affecting the release of the drug from the extrudates. Screening designs are commonly used when little is known about a system or process. These designs, in general, are fractional factorial of a  $2^n$  design that can identify main factors from a large number of suspected variables. Statistical tools such as Plackett–Burman design helps in selecting the most important variables (32). The Plackett–Burman method allows evaluation of ‘ $N-1$ ’ variables by ‘ $N$ ’ number of experiments ( $N$  must be a multiple

<sup>1</sup> Department Pharmaceutical Sciences and Technology, Institute of Chemical Technology, Matunga, Mumbai 400019, India.

<sup>2</sup> To whom correspondence should be addressed. (e-mail: dramin@vsnl.net)

of four). In the Plackett–Burman design, experiments are performed at various combinations of high and low values of the process variables and analyzed for their effect on the process (33,34). The Plackett–Burman design analyzes the input data and presents a rank ordering of the variables with magnitude of effect and designates signs to the effects to indicate whether an increase in factor values is advantageous or not (35).

Water solubility is one of the most important molecular properties of drugs for the development of extended release dosage forms as it is a key factor governing drug access to biological membranes (36). Therefore, in the present investigation, drugs having large difference in their solubility (theophylline and caffeine with solubility of 9.91 mg/mL and 136 mg/mL, respectively) was selected as model drugs. The objective of the study was to use the Plackett–Burman design to quantify the effect of amount and molecular weight of poly (ethylene oxide), amount and solubility of drug, ethylcellulose, sodium chloride and citric acid, polyethylene glycol and glycerin, amount on the mechanism, and rate of drug release from the extended release hot melt extrudates. In order to achieve the above-mentioned objective, a mathematical model that will work for a wide range of solubility of drugs will be developed and validated.

## MATERIAL AND METHODS

### Materials

Theophylline and caffeine were gifted by Bajaj Healthcare Ltd., India, poly (ethylene oxide) and ethylcellulose were gifted by Dow Chemical Company. All other chemicals and solvents used were of analytical grade and were procured from Merck India Ltd. Purified water was used throughout the study.

### Experimental Design

The Plackett–Burman factorial design was employed in this study to correlate dependent and independent variables using the following polynomial model:

$$Y = A_0 + A_1X_1 + A_2X_2 + A_3X_3 + \dots + A_nX_n$$

where  $Y$  is the response,  $A_0$  the constant and  $A_1$  to  $A_n$  are the coefficients of the response values.

The levels of independent and dependent variables evaluated in this study are listed in Table I. A nine-factor 12-run Plackett–Burman screening design was generated using Design-Expert® 6.0.10 (Version 2.05, Stat-Ease Inc., Minneapolis, USA; Table II). The software package was used to estimate the response of dependent variables and optimized conditions.

Each variable was represented at two levels, namely, “high” and “low”. These levels define the upper and lower limits of the range covered by each variable. In addition to the variables of real interest, the Plackett–Burman design considers insignificant dummy variables, whose number should be one-third of all variables. The dummy variables, which are not assigned any values, introduce some redundancy required by the statistical procedure. Incorporation of

**Table I.** Factors in the Plackett–Burman Screening Design

	Code	Low level	High level
Independent factors			
Poly (ethylene oxide) molecular weight	$X_1$	$6 \times 10^5$	$7 \times 10^6$
Poly (ethylene oxide) amount (mg)	$X_2$	100.00	300.00
Dummy	$X_3$	-1.00	1.00
Ethylcellulose amount (mg)	$X_4$	0.00	50.00
Drug solubility (mg/mL)	$X_5$	9.91	136.00
Drug amount (mg)	$X_6$	100.00	200.00
Sodium chloride amount (mg)	$X_7$	0.00	20.00
Citric acid amount (mg)	$X_8$	0.00	5.00
Polyethylene glycol amount (mg)	$X_9$	0.00	5.00
Dummy	$X_{10}$	-1.00	1.00
Glycerin amount (mg)	$X_{11}$	0.00	5.00
Dependent factors			
Time to release 90% of the drug	$Y_1$		
$n$ value	$Y_2$		
% amount released in 4 h	$Y_3$		
% amount released in 8 h	$Y_4$		

the dummy variables into an experiment allows an estimation of the variance (experimental error) of an effect.

### Hot Melt Extrusion

The composition of the hot melt extrudates was selected based on the statistical design presented in Table I. All the ingredients were passed through a #30 sieve and mixed in a blender for 10 min. The blend was fed into a single-screw hot melt extruder (fabricated by S.B. Panchal and Co., India) equipped with a 0.8-mm die. The screw speed was kept constant at 20 rpm. The temperature of the system was gradually increased till the extrusion process started and then it was kept constant at 80°C to 100°C. The extrudates were allowed to cool to room temperature and then were uniformly cut to the size of 5-mm length and filled into the size ‘0’ hard gelatin capsule shells such that each capsule contains 400 mg of extrudates.

### In Vitro Release Studies

*In vitro* release studies were performed using USP dissolution apparatus 1 at 100 rpm in 900 mL of phosphate buffer pH 6.8 (Electrolab India Ltd., India) at  $37 \pm 0.5^\circ\text{C}$ . Aliquots were withdrawn at predetermined time intervals, filtered and analyzed spectrophotometrically. All the studies were carried out in triplicates. T90 values were calculated by least square linear regression analysis.

### Release Exponent Estimation

Release exponent ( $n$ ) was estimated by fitting dissolution data to the Korsmeyer’s equation (37) as shown below.

$$\begin{aligned} Q_t/Q_\infty &= K.t^n \\ \log Q_t/Q_\infty &= \log K + n \log t \end{aligned}$$

**Table II.** Plackett–Burman Screening Design with Nine Variables Generated Using Stat-Ease Software

Formula	$X_1$	$X_2$	$X_3$	$X_4$	$X_5$	$X_6$	$X_7$	$X_8$	$X_9$	$X_{10}$	$X_{11}$
1	$7 \times 10^6$	100.00	1.00	0.00	9.91	100.00	20.00	5.00	5.00	-1.00	5.00
2	$7 \times 10^6$	300.00	-1.00	50.00	9.91	100.00	0.00	5.00	5.00	1.00	0.00
3	$6 \times 10^5$	300.00	1.00	0.00	136.0	100.00	0.00	0.00	5.00	1.00	5.00
4	$7 \times 10^6$	100.00	1.00	50.00	9.91	200.00	0.00	0.00	0.00	1.00	5.00
5	$7 \times 10^6$	300.00	-1.00	50.00	136.00	100.00	20.00	0.00	0.00	-1.00	5.00
6	$7 \times 10^6$	300.00	1.00	0.00	136.00	200.00	0.000	5.00	0.00	-1.00	0.00
7	$6 \times 10^5$	300.00	1.00	50.00	9.91	200.00	20.00	0.00	5.00	-1.00	0.00
8	$6 \times 10^5$	100.00	1.00	50.00	136.00	100.00	20.00	5.00	0.00	1.00	0.00
9	$6 \times 10^5$	100.00	-1.00	50.00	136.00	200.00	0.00	5.00	5.00	-1.00	5.00
10	$7 \times 10^6$	100.00	-1.00	0.00	136.00	200.00	20.00	0.00	5.00	1.00	0.00
11	$6 \times 10^5$	300.00	-1.00	0.00	9.91	200.00	20.00	5.00	0.00	1.00	5.00
12	$6 \times 10^5$	100.00	-1.00	0.00	9.91	100.00	0.00	5.00	0.00	-1.00	0.00

### Validation of the Model

The developed model was validated by conducting two additional experiments. The practical responses obtained were compared with those generated by the software.

## RESULT AND DISCUSSION

### Statistical Design and Analysis

The Plackett–Burman screening design was used to evaluate the effect of the nine independent variables on the release of the drug from the extrudates. Low and high values for each factor tested in screening design were identified in preliminary experiments. The magnitude of responses for each 12 experiments (observed and predicted) is given in Table III along with residual values. The observed and predicted values were found to be in close agreement with each other. All the residual values were found to be less than 1.5 for all the four responses which ensures the suitability of the model.

Polynomial equations were generated for all the responses which are listed in Table IV. The magnitude and direction of the factor coefficient in the all the four equations explains the nature of the effect of factors on the responses.

Factors with coefficients of greater magnitude show a high effect on the response suggesting that poly (ethylene oxide) amount, ethylcellulose amount, and solubility of drug demonstrated very significant effect on all four responses. Polyethylene glycol amount and glycerin amount also illustrated considerable effect on all four responses. The response value is directly proportional to the positive coefficients in the equations and inversely to the negative coefficients.

Using analysis of variance (ANOVA), the significance ( $F$  value  $< 0.05$ ) of the ratio of mean square variation due to regression coefficient and residual error was tested. The regression coefficient and probability values obtained for all the four responses were as shown in Table V. The value of more than 0.99 for regression coefficient and less than 0.05 for probability indicated the significance of the model except for the response  $Y_3$  (% of drug release in 4 h). The analysis of variance for all the four responses was as shown in the Tables VI, VII, VIII, and IX.

### Effect of Individual Factor in Presence of Other Factors on Drug Release and Release Exponent

Dissolution profiles of the formulations 1–6 and 7–12 were as shown in the Figs. 1 and 2 respectively. Time required for 90% of drug released ( $T_{90}$ ) ranged between 2.75 h

**Table III.** Observed and Predicted Values of the Responses

No.	$Y_1$			$Y_2$			$Y_3$			$Y_4$		
	OV	PV	R	OV	PV	R	OV	PV	R	OV	PV	R
1	3.49	3.28	0.21	0.100	0.10	0.00	94.63	94.73	-0.1	97.12	97.25	-0.13
2	11.09	11.30	-0.21	0.79	0.81	-0.02	62.03	60.88	1.15	80.67	80.54	0.13
3	3.93	3.94	-0.01	0.33	0.35	-0.02	91.85	91.95	-0.1	99.29	98.61	0.68
4	5.72	5.73	-0.01	0.34	0.34	0.00	86.90	88.05	-1.15	93.37	92.69	0.68
5	7.87	7.86	0.01	0.75	0.73	0.02	67.94	66.79	1.15	88.92	89.60	-0.68
6	3.88	3.67	0.21	0.26	0.24	0.02	87.18	88.33	-1.15	99.54	99.67	-0.13
7	14.18	13.97	0.21	0.94	0.94	0.00	43.54	44.69	-1.15	73.64	73.77	-0.13
8	4.06	4.07	-0.01	0.41	0.41	0.00	88.78	88.88	-0.1	99.33	98.65	0.68
9	3.13	3.12	0.01	0.10	0.11	-0.01	94.20	94.10	0.1	99.07	99.75	-0.68
10	4.24	4.45	-0.21	0.09	0.10	-0.01	92.34	92.24	0.1	95.03	94.90	0.13
11	4.94	5.15	-0.21	0.40	0.40	0.00	89.34	89.24	0.1	95.39	95.26	0.13
12	2.75	2.74	0.01	0.30	0.28	0.02	94.55	93.40	1.15	99.94	100.62	-0.68

OV observed value, PV predicted value, R residual

**Table IV.** Regression Equations of the Fitted Models

$$Y_1 = 5.77 + 0.28X_1 + 1.88X_2 + 1.90X_4 - 1.25X_5 + 0.24X_6 + 0.69X_7 - 0.67X_8 + 0.90X_9 - 0.93X_{11}$$

$$Y_2 = 0.40 - 0.013X_1 + 0.18X_2 + 0.15X_4 - 0.078X_5 - 0.045X_6 + 0.047X_7 - 0.057X_8 - 0.0085X_9 - 0.064X_{11}$$

$$Y_3 = 82.77 - 0.94X_1 - 9.13X_2 - 8.88X_4 + 4.27X_5 - 0.52X_6 - 3.34X_7 + 3.25X_8 - 3.01X_9 + 4.70X_{11}$$

$$Y_4 = 93.44 - 1.00X_1 - 3.87X_2 - 4.28X_4 + 3.42X_5 - 0.77X_6 - 1.87X_7 + 1.74X_8 - 2.64X_9 + 2.08X_{11}$$

(formulation 12) and 14.18 h (formulation 7). Release exponent ( $n$ ) values varied from 0.1 (formulation 1) and 0.94 (formulation 7). The interrelationships of the three major parameters poly (ethylene oxide) amount, ethylcellulose amount and solubility of drug) in presence of middle level of remaining parameters are illustrated in Figs. 3, 4, and 5.

In order to study the behavior of the individual parameters in the presence of other parameters, each parameter was treated alternatively at the highest and lowest level while keeping all other remaining parameters constant at their middle levels in equations given in Table IV. The effect of the individual independent variables in the presence of the middle level of the remaining factors is described in Table X and is further discussed in the following sections. Factors having positive coefficient (in equations of Table IV) for  $Y_1$  and negative coefficient for  $Y_3$  and  $Y_4$  will be the ones which will help in retarding the release of the drug and vice versa. Response  $Y_2$  is an indicator of release mechanism and was calculated using Korsmeyer's equation. To confirm the diffusion mechanism, the data were fit to the Korsmeyer's equation (37);

$$Qt/Q\alpha = K \cdot t^n$$

where  $Qt$  is the amount released at time  $t$ ,  $Q\alpha$  is overall released amount,  $K$  is a constant incorporating the properties of macromolecular polymeric system and the drug, and  $n$  is a kinetic constant that depends on the transport mechanism. The exponent  $n$  gives information about the release mechanism; whereby, a value of  $n < 0.5$  illustrates Fickian diffusion,  $n = 0.5$  characterizes diffusion-controlled release,  $0.5 < n < 1.0$  indicates anomalous (non-Fickian transport), and  $n = 1.0$  indicates swelling controlled release (zero order kinetics). Drug diffusion and polymer erosion control the release process in equal parts, if  $n = 0.66$  (38).

**Table V.** Probability and Correlation Coefficient Values for the Responses

Factors	Responses			
	$Y_1$	$Y_2$	$Y_3$	$Y_4$
Prob>F	0.0089	0.0045	0.1127	0.0164
Regression	0.9980	0.9990	0.9738	0.9963

**Table VI.** Analysis of Variance for Response  $Y_1$

Source	Sum of squares	df	Mean square	F value	Prob>F
$X_1$	0.91	1	0.91	6.64	0.1233
$X_2$	42.19	1	42.19	308.69	0.0032
$X_4$	43.40	1	43.40	317.53	0.0031
$X_5$	18.90	1	18.90	138.29	0.0072
$X_6$	0.70	1	0.70	5.13	0.1518
$X_7$	5.71	1	5.71	41.80	0.0231
$X_8$	5.47	1	5.47	40.01	0.0241
$X_9$	9.79	1	9.79	71.65	0.0137
$X_{11}$	10.30	1	10.30	75.40	0.0130
Residual	0.27	2	0.14		
Total (corrected)	137.64	11			

Standard deviation of the residuals = 0.37. Explained variation about the mean = 99.80

*Effect of Poly (ethylene oxide) Molecular Weight ( $X_1$ )*

As suggested by the polynomial equations in Table IV, molecular weight of poly (ethylene oxide) had a very insignificant effect on all the four responses. The effect of poly (ethylene oxide) molecular weight on  $Y_1$ , at mid level of the remaining factors can be seen from Table X. Increasing poly (ethylene oxide) molecular weight from  $6 \times 10^5$  to  $7 \times 10^6$  resulted in increasing T90 value from 5.5 to 6.05 h indicating that poly (ethylene oxide) molecular weight had slight effect on retarding the release of drug. There was no change in the release mechanism with  $n$  value decreasing from 0.41 (Fickian diffusion) to 0.39 (Fickian diffusion) with increase in molecular weight.

*Effect of Poly (ethylene oxide) Amount ( $X_2$ )*

Having the second highest coefficient of +1.88, poly (ethylene oxide) amount exhibited a very significant effect on all the four responses. About two times decrease in T90 value was observed when the poly (ethylene oxide) amount was increased from its lower value of 100 mg to a higher value of 300 mg. As the amount of poly (ethylene oxide) increased,

**Table VII.** Analysis of Variance for Response  $Y_2$

Source	Sum of squares	df	Mean square	F value	Prob>F
$X_1$	0.00218	1	0.00219	5.02	0.1543
$X_2$	0.39	1	0.38	861.96	0.0012
$X_4$	0.29	1	0.29	656.57	0.0015
$X_5$	0.073	1	0.073	166.99	0.0059
$X_6$	0.025	1	0.025	56.65	0.0172
$X_7$	0.027	1	0.027	60.89	0.0160
$X_8$	0.039	1	0.039	89.04	0.0110
$X_9$	0.000867	1	0.00086	1.99	0.2936
$X_{11}$	0.049	1	0.049	112.32	0.0088
Residual	0.00087	2	0.00043		
Total (corrected)	0.88	11			

Standard deviation of the residuals = 0.021. Explained variation about the mean = 99.90

**Table VIII.** Analysis of Variance for Response  $Y_3$ 

Source	Sum of squares	df	Mean square	F value	Prob>F
$X_1$	10.53	1	10.53	0.28	0.6507
$X_2$	999.55	1	999.55	26.39	0.0359
$X_4$	945.19	1	945.19	24.95	0.0378
$X_5$	219.31	1	219.31	5.79	0.1379
$X_6$	3.29	1	3.29	0.087	0.7961
$X_7$	134.27	1	134.27	3.54	0.2005
$X_8$	127.01	1	127.01	3.35	0.2086
$X_9$	108.60	1	108.60	2.87	0.2325
$X_{11}$	265.46	1	265.46	7.01	0.1180
Residual	75.77	2	37.88		
Total (corrected)	2,888.96	11			

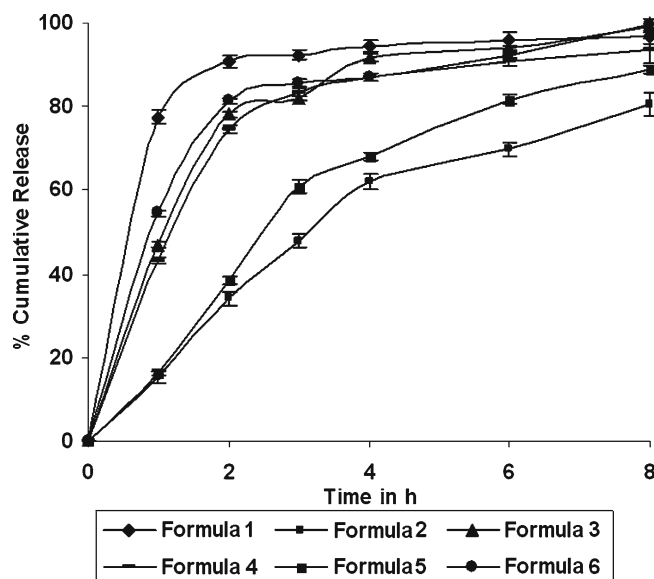
the number of entangling polymer chains and consequent entrapment of the drug inside the polymer network increased, which cause a delay in drug release. Similar effects were also observed for response  $Y_3$  and  $Y_4$ , where in the amount of drug released was decreased with increase in poly (ethylene oxide) amount. The amount of poly (ethylene oxide) also had a significant effect on the release exponent ( $n$ ). Increasing the amount of poly (ethylene oxide) from 100 mg to 300 mg increased the  $n$  values from 0.22 to 0.58. Increasing the amount of poly (ethylene oxide) resulted in decreasing the fluid-filled channels through which the drug may diffuse and increasing the transpolymer diffusional pathway of matrix. Fluid-filled channels are characteristic of the Fickian square root of time release pattern whereas transpolymer diffusion is pertinent to anomalous drug release where  $n > 0.5$  (39).

To study the effect of poly (ethylene oxide) amount in presence of other two significant factors ethylcellulose amount and drug solubility, a three-dimensional surface graph was constructed (Figs. 3 and 4). Increasing poly (ethylene oxide) and ethylcellulose amount increased response  $Y_1$ . Thus, both these factors act together in harmony to retard the release of the drug (Fig. 3). However, drug solubility act in a reverse manner as that of poly (ethylene oxide) and ethylcellulose amount, thereby by enhancing the release of drug with increase in its solubility.

**Table IX.** Analysis of Variance for Response  $Y_4$ 

Source	Sum of squares	df	Mean square	F value	Prob > F
$X_1$	12.02	1	12.02	8.43	0.1010
$X_2$	179.49	1	179.49	125.90	0.0078
$X_4$	219.39	1	219.39	153.89	0.0064
$X_5$	140.43	1	140.43	98.50	0.0100
$X_6$	7.10	1	7.10	4.98	0.1553
$X_7$	42.00	1	42.00	29.46	0.0323
$X_8$	36.51	1	36.51	25.61	0.0369
$X_9$	83.58	1	83.58	58.63	0.0166
$X_{11}$	52.13	1	52.13	36.56	0.0263
Residual	185.927	2	1.43		
Total (corrected)	5,279.80	11			

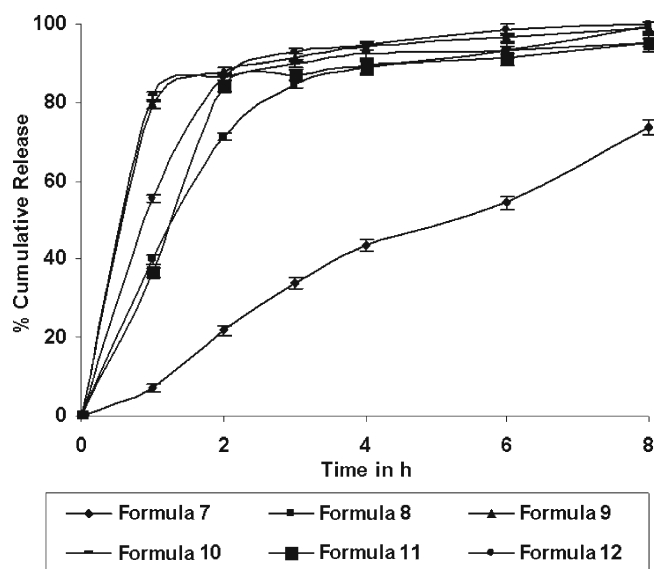
Standard deviation of the residuals = 1.19. Explained variation about the mean = 99.63

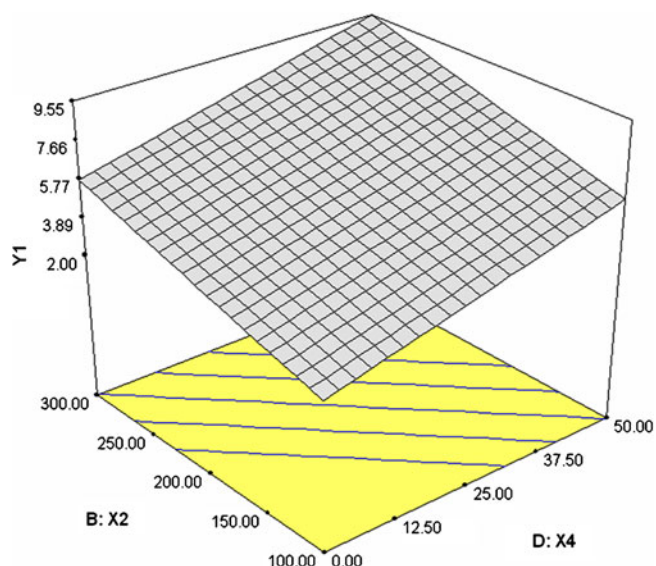
**Fig. 1.** Release profile of the formulation 1 to formulation 6

#### Effect of Ethylcellulose Amount ( $X_4$ )

Ethylcellulose is an inert and hydrophobic polymer prepared by treating purified cellulose with an alkaline solution followed by ethylation of the alkali cellulose with chloroethane (40). It has been used as matrix-forming material for sustained release dosage forms (41, 42). The mechanism of drug release from the ethylcellulose matrix is simple diffusion for water-soluble drugs and diffusion followed by polymer relaxation for slightly soluble and practically insoluble drugs (43). Ethylcellulose, being a hydrophobic matrix, further retards release of the drugs from the matrix and was found to be the major controlling factor for the system. When concentration of ethylcellulose was increased from 0 to 50 mg; T90 values were increased from 3.87 to 7.68 h at medium levels of all the other variables.

It also affected mechanism of drug release. At medium level of all the other factors, as concentration of ethylcellulose

**Fig. 2.** Release profile of the formulation 7 to formulation 12



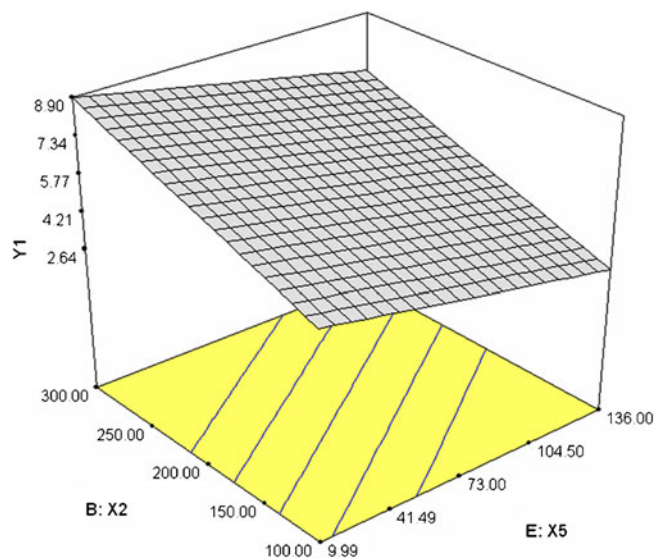
**Fig. 3.** 3-D plot illustrating the effect of poly (ethylene oxide) amount ( $X_2$ ) and ethylcellulose amount ( $X_4$ ) on response  $Y_1$

was increased from 0 mg to 50 mg,  $n$  value increased from 0.25 to 0.56. Thus, with increase in ethylcellulose amount, the mechanism of drug release changed from Fickian diffusion to anomalous. These results are in agreement with the reported literature (43).

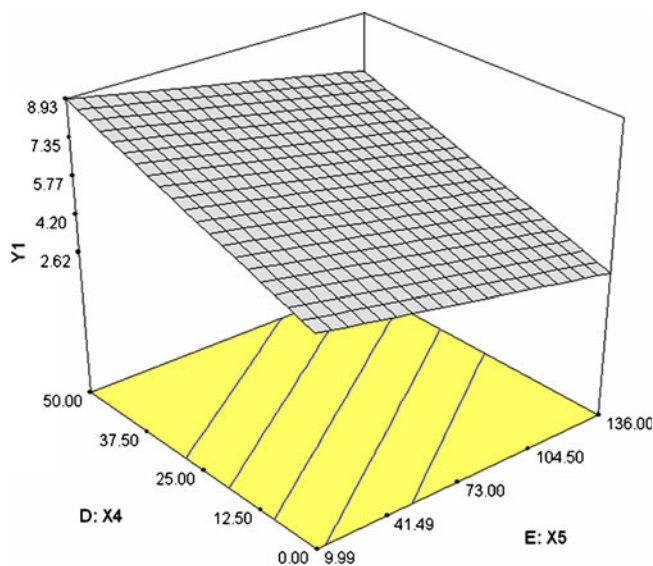
To study the effect of ethylcellulose amount in the presence of two other significant factors, poly (ethylene oxide) amount and drug solubility, a three-dimensional surface graph was constructed (Figs. 3 and 5). As discussed above, ethylcellulose amount and poly (ethylene oxide) amount together helped in reducing the release of the drug and thereby increasing the T90 value

#### Effect of Drug Solubility ( $X_5$ )

Drug solubility had significant effect on the release of the drug from the matrix. However, unlike the above three



**Fig. 4.** 3-D plot illustrating the effect of poly (ethylene oxide) amount ( $X_2$ ) and drug solubility ( $X_5$ ) on response  $Y_1$



**Fig. 5.** 3-D plot illustrating the effect of ethylcellulose amount ( $X_4$ ) and drug solubility ( $X_5$ ) on response  $Y_1$

parameters discussed, drug solubility with negative coefficient in the polynomial equation is inversely related to T90 value and helps in enhancing the release of the drug. At medium level of all the other factors, T90 value varied from 7.03 to 4.52 h with increase in drug solubility from 9.91 mg/mL to 136 mg/mL. This was in accordance with the literature (44,45). Drug solubility also contributed towards the release mechanism by decreasing the  $n$  value from 0.50 to 0.32 changing the mechanism from diffusion-controlled release to Fickian diffusion.

#### Effect of Drug Amount ( $X_6$ )

The T90 value changes from 5.53 to 6.01 h when drug amount was changed from 100 mg to 200 mg. The data itself indicate the minimal effect of drug amount on the  $Y_1$ . Similarly, it showed minimum effect on remaining three responses.

#### Effect of Sodium Chloride ( $X_7$ )

Sodium chloride showed positive effect on the T90 values, thereby reducing the release rate. Its effect on the  $Y_1$  and  $Y_2$  was found to be insignificant. An increase in amount of sodium chloride creates more osmotic pressure difference and pulls more water into the dosage form. In the presence of more water, poly (ethylene oxide) forms a highly viscous gel and subsequently controls the drug release. The results are in agreement with those reported by Sastry *et al.* (46).

#### Effect of Citric Acid ( $X_8$ )

Citric acid was added to the formulation so as to modulate the pH of the microenvironment, which would have an effect on the drug release. Data revealed that citric acid has insignificant effect on all responses. Apart from modulating micro environmental pH, increased drug release is due to its solubility in water and thereby its ability to form channels within the polymer matrix.

**Table X.** Response Values at High and Low Level of Individual Factor at Middle Level of Remaining Factors

Sr. No.	Factor	Code	Level	Response Value at middle level of other factors			
				Y <sub>1</sub>	Y <sub>2</sub>	Y <sub>3</sub>	Y <sub>4</sub>
1	Poly (ethylene oxide)	X <sub>1</sub>	Low	5.5	0.41	83.71	94.44
	molecular weight	X <sub>1</sub>	High	6.05	0.39	81.84	92.44
2	Poly (ethylene oxide)	X <sub>2</sub>	Low	3.9	0.22	91.9	97.31
	amount (mg)	X <sub>2</sub>	High	7.65	0.58	73.65	89.57
3	Ethylcellulose	X <sub>1</sub>	Low	3.87	0.25	91.65	97.71
	amount (mg)	X <sub>1</sub>	High	7.68	0.56	73.9	89.16
4	Drug solubility	X <sub>2</sub>	Low	7.03	0.50	78.5	90.02
	(mg/mL)	X <sub>2</sub>	High	4.52	0.32	87.04	96.86
5	Drug amount (mg)	X <sub>1</sub>	Low	5.53	0.45	83.3	94.21
		X <sub>1</sub>	High	6.01	0.36	82.25	92.67
6	Sodium chloride	X <sub>2</sub>	Low	5.08	0.35	86.11	95.31
	amount (mg)	X <sub>2</sub>	High	6.46	0.45	79.43	91.57
7	Citric acid amount	X <sub>1</sub>	Low	6.45	0.46	79.52	91.69
	(mg)	X <sub>1</sub>	High	5.1	0.34	86.02	95.18
8	Polyethylene glycol	X <sub>2</sub>	Low	4.87	0.41	85.78	96.08
	amount (mg)	X <sub>2</sub>	High	6.68	0.39	79.77	90.8
9	Glycerin amount	X <sub>2</sub>	Low	6.7	0.46	78.07	91.35
	(mg)	X <sub>2</sub>	High	4.85	0.34	87.48	95.52

X<sub>1</sub>–Poly (ethylene oxide) molecular weight

X<sub>2</sub>–Poly (ethylene oxide) amount

#### Effect of Polyethylene Glycol (X<sub>9</sub>)

Polyethylene glycol was incorporated in the formulation as plasticizer. Plasticizers are used to increase the flexibility of the polymer. At medium levels of the all the other variables, as the concentration of polyethylene glycol was increased from 0 to 5 mg, the T90 value changes from 4.87 to 6.68 h. Also, the *n* value was changed from 0.41 to 0.39 suggesting that there was no change in the release mechanism.

#### Effect of Glycerin (X<sub>11</sub>)

At medium levels of all the other variables, as the concentration of glycerin was increased from 0 to 5 mg, the T90 value changes from 6.70 to 4.85 h. This was in accordance to the literature (47). Also, the *n* value was changed from 0.46 to 0.34. From the data, it is clear that the effect of glycerin and polyethylene glycol was of similar magnitude but in opposite directions.

#### Validation of the Model

The formulations used for the development of the validation batches were as shown in the Table XI. The results of the developed formulations were as shown in Table XII. The difference between the predicted and experimental values was within the limits of  $\pm 1.0$ , as required by the model.

#### CONCLUSION

Within the limits commonly used to prepare controlled release matrices by melt extrusion technology, some formulation variables are expected to have significant effect on the amount and pattern of drug release. Poly (ethylene oxide) amount, ethylcellulose amount and drug solubility had significant effect on the T90 values whereas poly (ethylene oxide) amount and ethylcellulose amount had a significant effect on the mechanism of release. From this study it can be concluded that sustained release matrices can be prepared easily by combining ethylcellulose and poly (ethylene oxide). Ethylcellulose plays a major role in controlling the drug release for longer duration. Poly (ethylene oxide) content is expected to have an impact on the pattern by which the drug is released from the matrix. Poly (ethylene oxide) amount, drug loading, electrolytes, buffers and plasticizers on the

**Table XI.** Composition of the Validation Batches

Formula	Quantities in mg	
	Formulation 1	Formulation 2
Ingredients		
Poly (ethylene oxide)		
WSR 303 (molecular weight $7 \times 10^6$ )	292.21	298.86
Ethylcellulose N-7	47.66	47.50
Theophylline	134.70	125.97
Sodium chloride	19.87	15.86
Citric acid	0.36	0.12
Polyethylene Glycol	4.08	4.94
Glycerin	0.28	0.81
Total	499.16	494.06

**Table XII.** Predicted and Observed Value of the Validation Batches

Response	Formulation 1		Formulation 2	
	Predicted	Experimental	Predicted	Experimental
Y <sub>1</sub>	13.33	13.74	13.34	14.26
Y <sub>2</sub>	0.93	0.97	0.94	0.92
Y <sub>3</sub>	47.06	47.59	47.08	46.52
Y <sub>4</sub>	74.95	74.83	74.98	74.59

other hand are expected to have marginal effect on both the responses, assuming they were within the limits used in this study.

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